



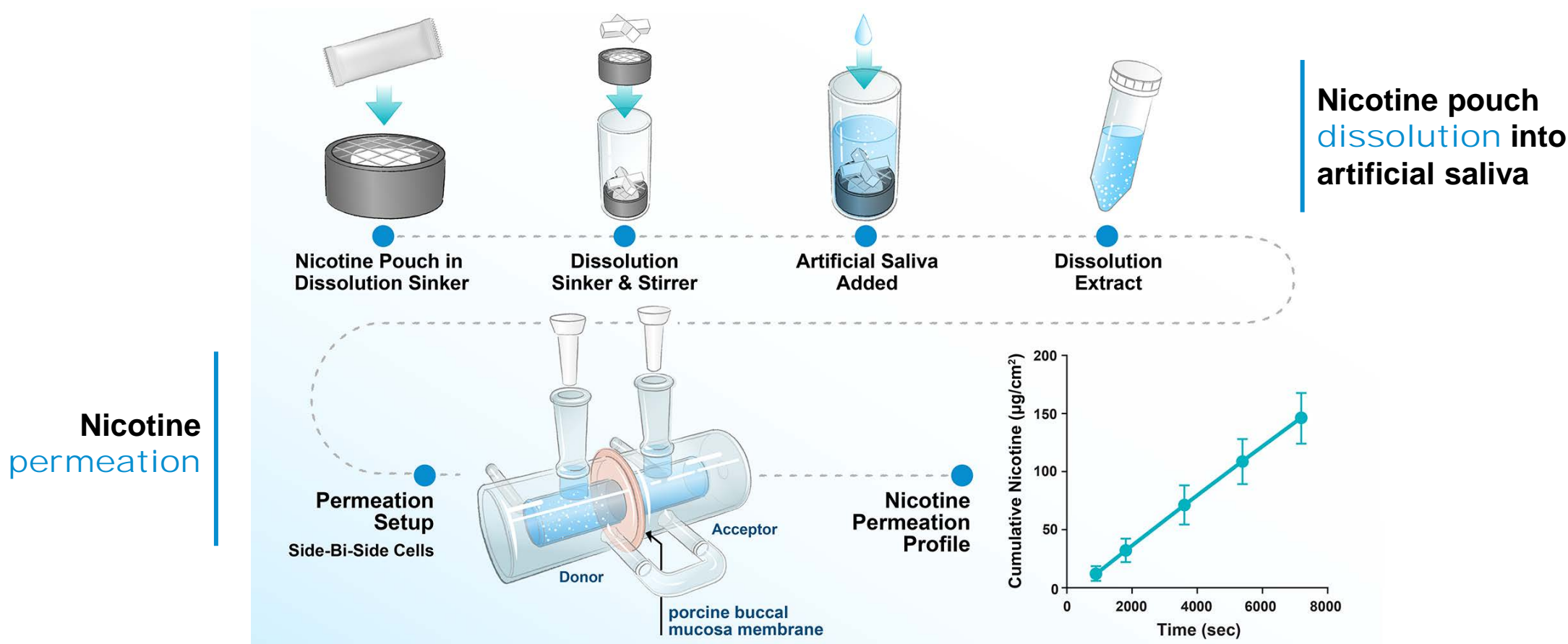
# Evaluation of Factors Impacting Nicotine Permeation from Oral Pouch Products

**Xia “Summer” Li and Akchara Sriram**



# Objective

Establish a screening tool to evaluate product attributes influencing nicotine release and buccal permeation from pouch products, supporting evaluation of product performance and formulation consistency.



# Chemical Characterization of Nicotine Pouch Products






- ✓ Nicotine
- ✓ pH
- ✓ Dissolution
  - USP-4
  - Sinkers

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# Nicotine Content and pH Values

Product	Nicotine Content (mg/pouch) (SD)	pH (SD)
 <b>Zyn<sup>®</sup> 6 mg Cool Mint</b>	5.58 (0.31)	8.22 (0.02)
 <b>Dryft<sup>®</sup> 7 mg Spearmint</b>	5.94 (0.50)	8.32 (0.05)
 <b>Velo<sup>®</sup> 4 mg Mint</b>	4.16 (0.24)	7.40 (0.02)

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# Dissolution Testing in Various Artificial Saliva

Product	Saliva		
	Buffer	Salivea	Orthana
<b>Zyn 6 mg Cool Mint</b>	✓	✓	✓
<b>Dryft 7 mg Spearmint</b>	—	—	✓
<b>Velo 4 mg Mint</b>	—	—	✓
<b>Saliva Composition</b>	Sodium Chloride Potassium Chloride Calcium Chloride Potassium Phosphate dibasic (anhydrous) Magnesium Chloride Hexahydrate Potassium Carbonate (anhydrous)  <small>*53160-1, D.V.T.M., Colorfastness to Saliva; Determination of the Colorfastness of Articles in Common Use Part 1: Resistance to Artificial Saliva. DIN V Test Method, 2022.</small>	Propylene Glycol Xylitol Hydrogenated Starch Hydrolysate Poloxamer 407 Hydroxyethyl Cellulose Sodium Benzoate Benzoic Acid Disodium Phosphate Zinc Gluconate Aloe Vera 7 Salivary Enzymes Potassium Chloride Sodium Chloride Sodium Bicarbonate	Porcine Gastric Mucin Methyl-4-hydroxybenzoate Benzalkonium Chloride EDTA Disodium Salt Potassium Hydroxide Xylitol Sodium Chloride Potassium Chloride Calcium Chloride Potassium Carbonate (anhydrous)

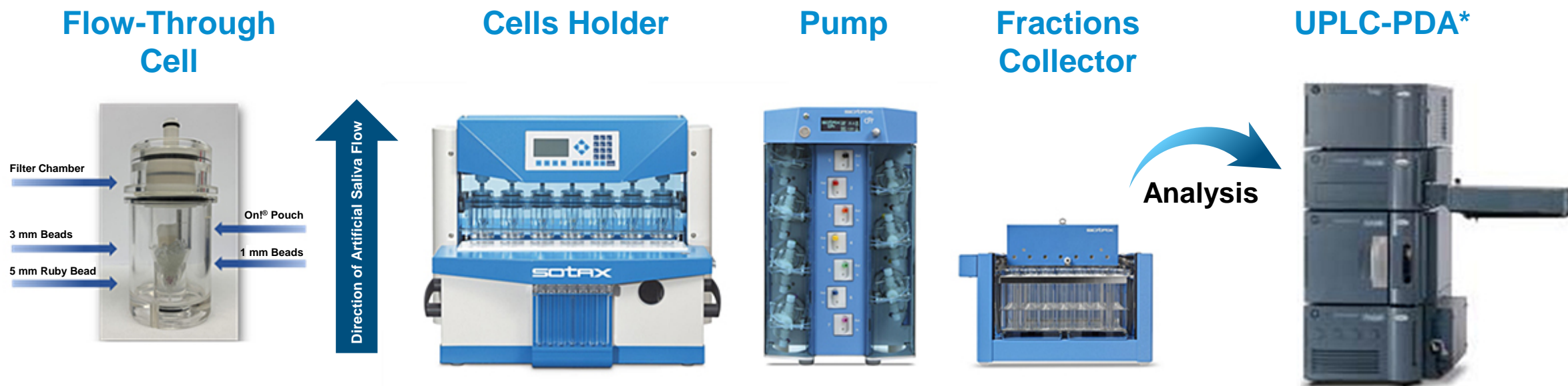
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# USP-4 Dissolution Methodology



**Flow rate 4.0 mL/min | Temperature 37° C | 12 replicates | 9 fractions per replicate**

\*UPLC-PDA: Ultra-high Performance Liquide Chromatography (UPLC)-Photodiode Array (PDA)

J. H. Miller, T. Danielson, Y. B. Pithawalla, A. P. Brown, C. Wilkinson, K. Wagner, and F. Aldeek. Development and validation of dissolution testing for nicotine release from smokeless tobacco products using flow-through cell apparatus and UPLC-PDA. *J. Chromatogr B* 2020, 1141, 122012.

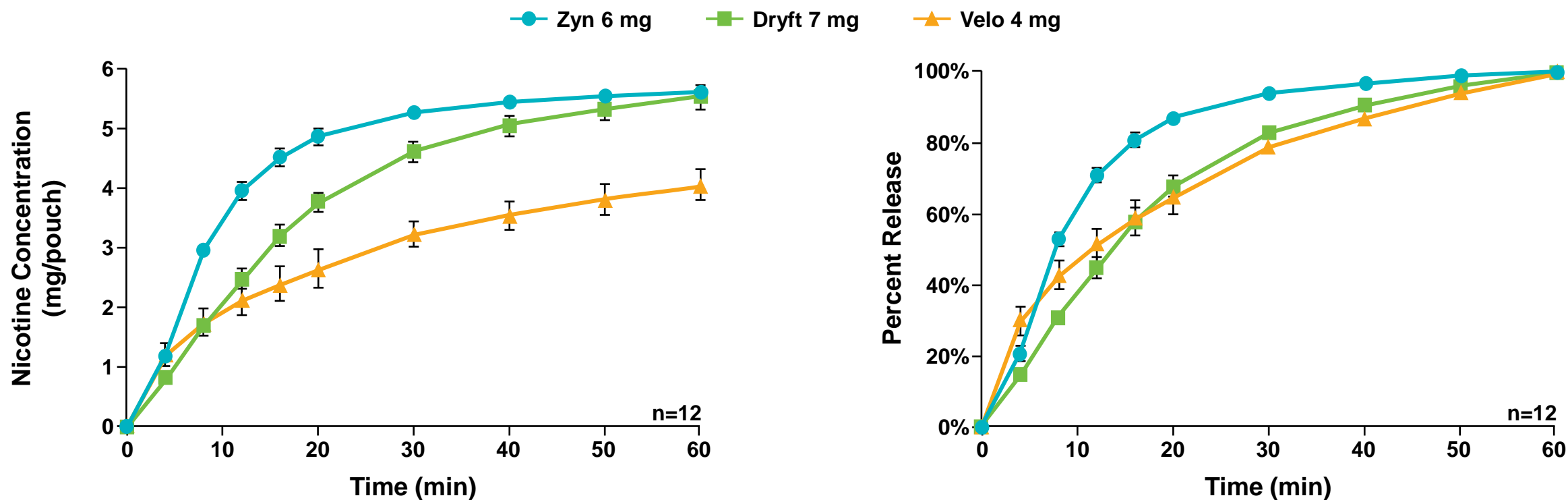
F. Aldeek, N. McCutcheon, C. Smith, J. H. Miller, and T. Danielson. Dissolution testing of nicotine release from OTDN pouches: product assessment and product-to-product comparison. *Separations*. 2021, 8 (1), 7.



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# USP-4 Nicotine Dissolution Profiles in Buffer Saliva



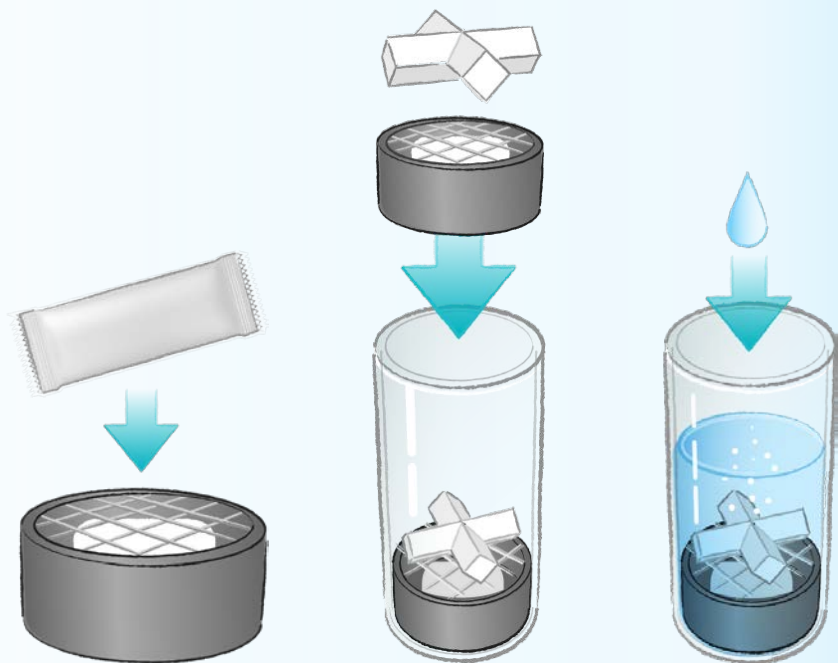
The total amount of nicotine released after 60 minutes of dissolution in buffer saliva was found to be 100%, 93% and 98% for *Zyn*, *Dryft* and *Velo*, respectively

*Velo* and *Dryft* exhibited equivalent release rates, both of which were slower than *Zyn*. This could be attributed to differences in the inherent product characteristics.



# Sinker Dissolution Methodology

## Low Volume Dissolution Setup



**Rotational Speed =**

► **100 rpm**

**Volume of Artificial Saliva =**

► **10 mL**

**Sampling Volume =**

► **250 µL**

**Sampling Time Points =**

► **4, 8, 12, 16, 20, 30,  
40, 50 and 60 minutes**

Knopp, M. M.; Kiil-Nielsen, N. K.; Masser, A. E.; Staaf, M., Introducing a Novel Biorelevant In Vitro Dissolution Method for the Assessment of Nicotine Release from Oral Tobacco-Derived Nicotine (OTDN) and Snus Products. Separations 2022, 9 (2), 52.

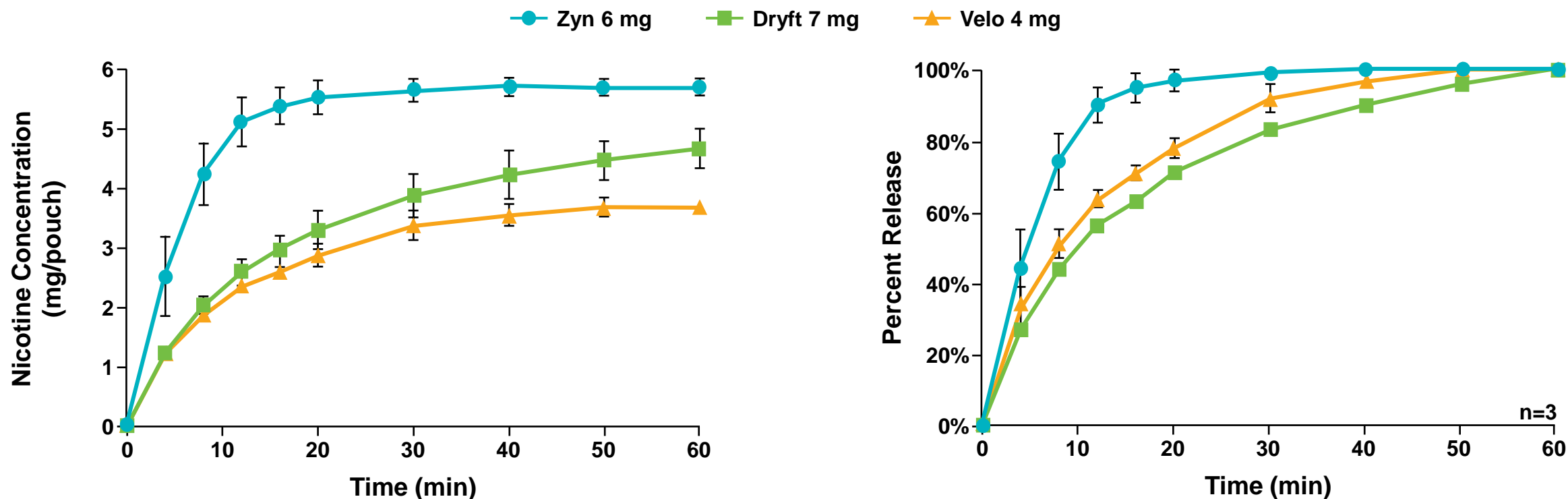


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# Sinker Method Dissolution Profiles in Orthana Saliva

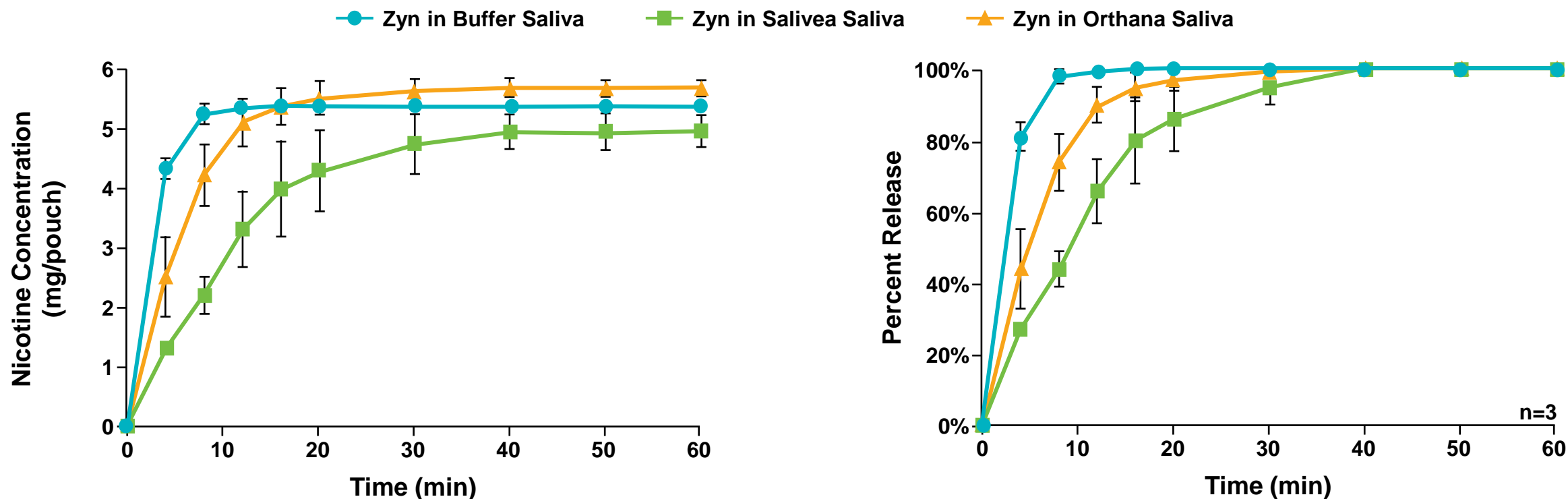


The total amount of nicotine released after 60 minutes of dissolution in orthana saliva, was found to be 102%, 79% and 89% for *Zyn*, *Dryft* and *Velo*, respectively

Release rates of the three nicotine pouches showed similar trend to USP-4, despite differences artificial saliva composition



# Influence of Saliva on Nicotine Release Rate



The rate of nicotine release was found to increase in the following order:  
**Salivea saliva < Orthana saliva < Buffer saliva**





# Conclusion



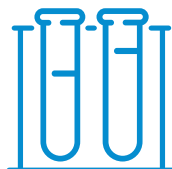
Differences in release rates  
may be due to inherent product characteristics

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Nicotine dissolution trends were similar  
across USP-4 and sinker methods

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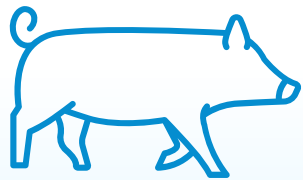


Saliva type  
affects nicotine release rate

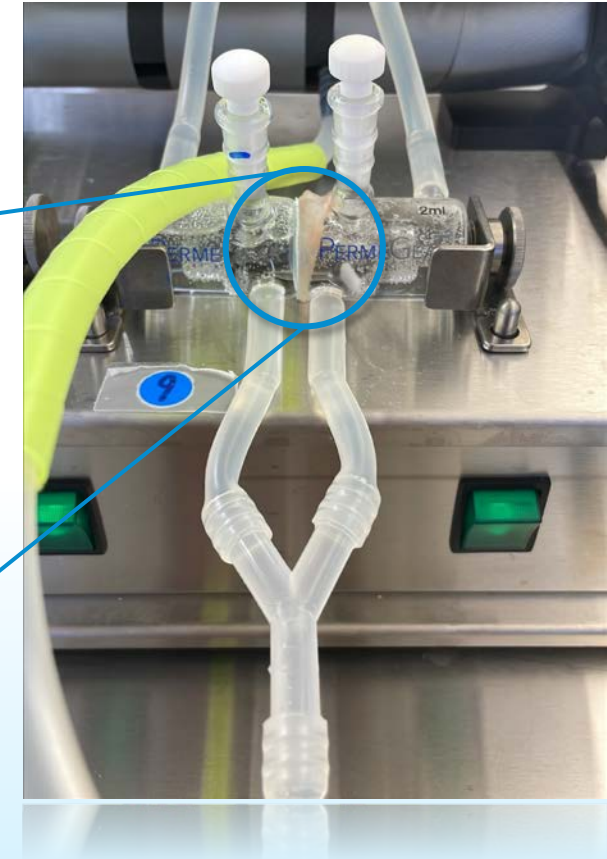


# Buccal Permeation Test

Test the delivery of active compound through the oral membrane



**Use porcine  
oral membrane**  
to emulate human  
oral membrane

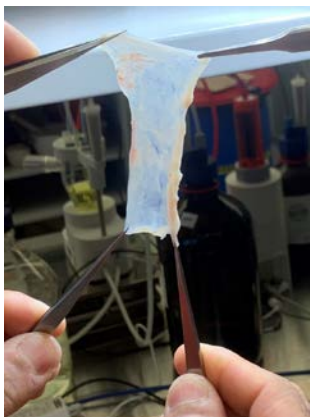


**Useful tool for formulation development of certain oral products, such as nicotine pouches**



# Diagram of *Ex Vivo* Permeation Experiment

**Tissue Preparation**



**Permeation Setup**



**Sample Collection**



**Sample Analysis**



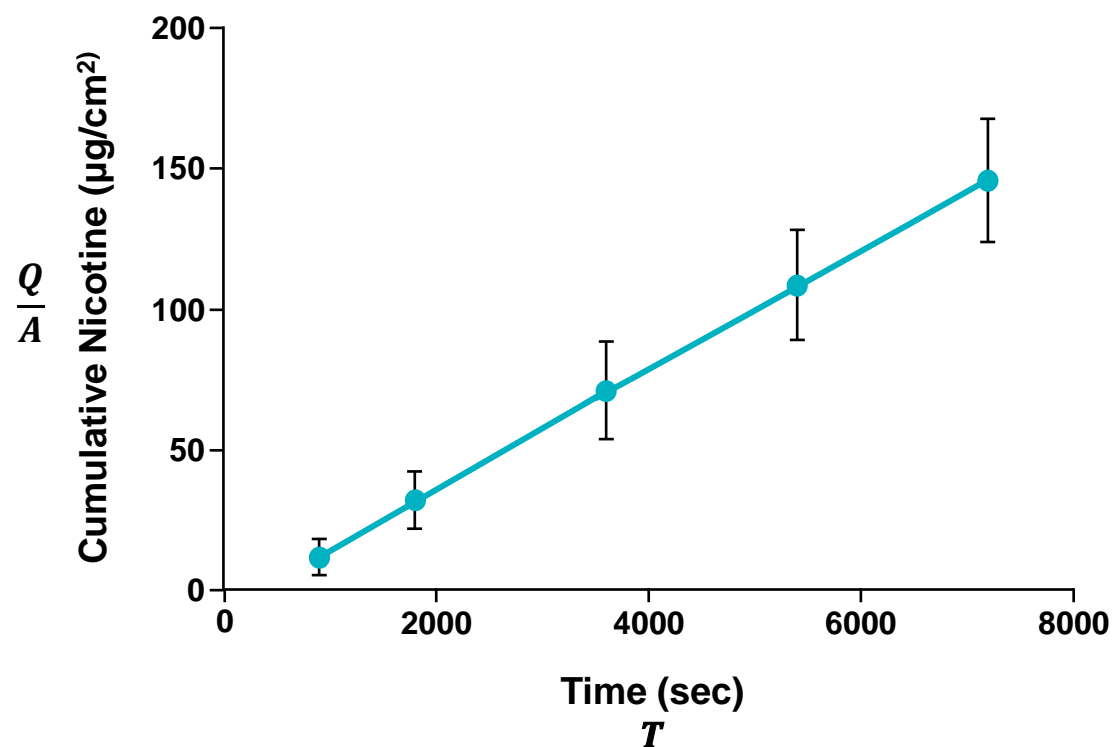
**Permeation  
Experiment Duration:  
2 hrs**

**Sample collections at:  
15, 30, 60, 90  
and 120 min**

**Ultra-Performance Liquid  
Chromatography-Mass  
Spectrometry (UPLC-MS)**



# Introduction to Apparent Permeability Coefficient ( $P_{app}$ )



Q: Cumulative amount of permeated drug

T: Time interval (t)

Flux: Slope of the linear regression of the cumulative plot of the drug through the barrier normalized by the surface area (A)

Apparent permeability coefficient ( $P_{app}$ ):  
Flux divided by concentration of the drug in the donor compartment at  $T_0$  ( $C_0$ )

$$Flux = \frac{dQ}{dt * A} \quad P_{app} = \frac{Flux}{C_0}$$



# Dissolution Samples with Different Nicotine Concentrations, pH, and Artificial Saliva

## Dissolution Samples of Zyn 6 mg Cool Mint in Various Artificial Saliva

Sample Name	Extract pH	Conc. of Nicotine (mg/mL)
Buffer saliva	8.06	0.60
Salivea saliva	8.04	0.48
Orthana saliva	7.79	0.60

## Dissolution Samples of Commercial Products in Orthana Saliva

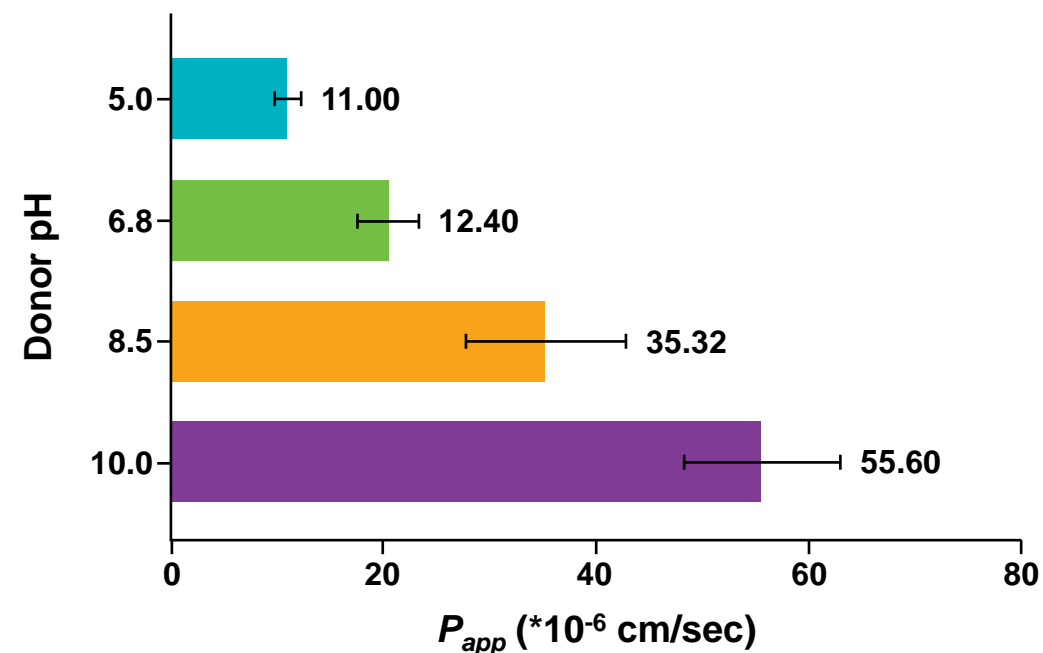
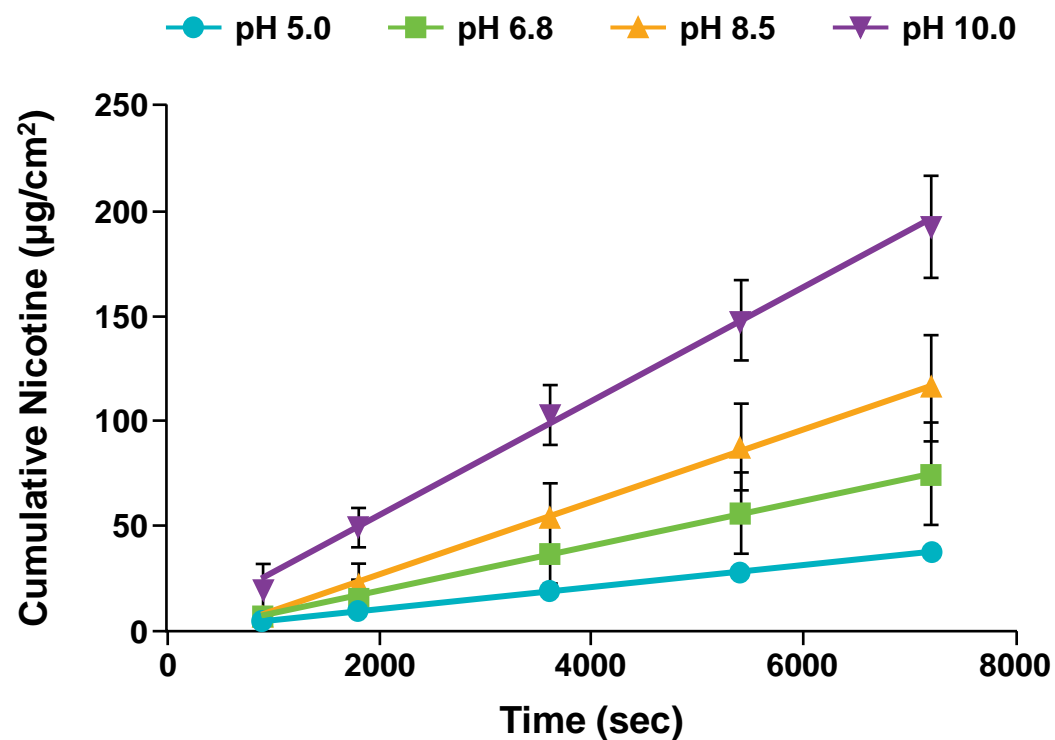
Sample Name	Extract pH	Conc. of Nicotine (mg/mL)
<i>Zyn 6 mg Cool Mint</i>	7.79	0.60
<i>Dryft 7 mg Spearmint</i>	7.74	0.45
<i>Velo 4 mg Mint</i>	6.94	0.38

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# Nicotine Permeability Increases as Donor pH Increases

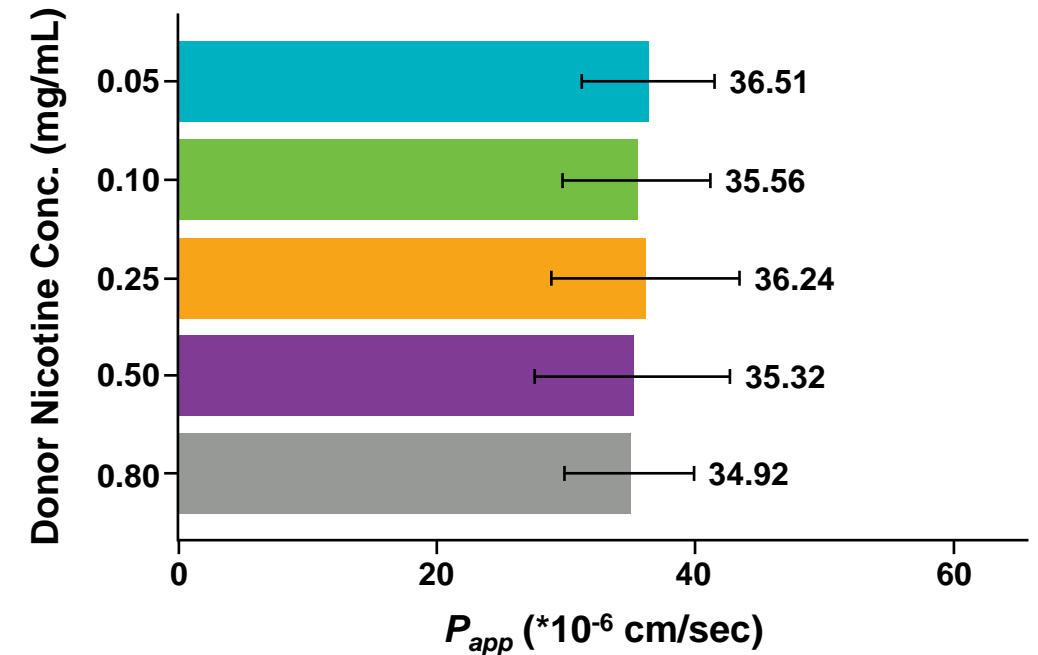
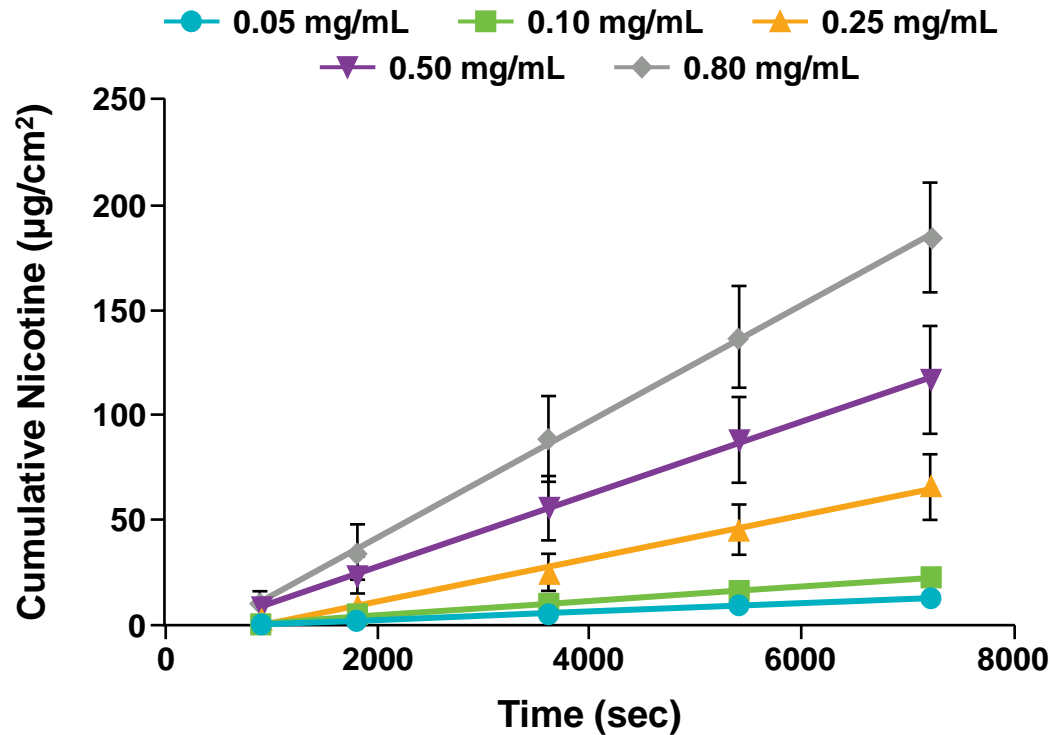
Donor: 0.5mg/mL Neat Nicotine Solution in PBS, n=12





# Concentration Does Not Impact Nicotine Permeability within the Tested Range

Donor: Neat Nicotine Solution in PBS, pH 8.5, n=12

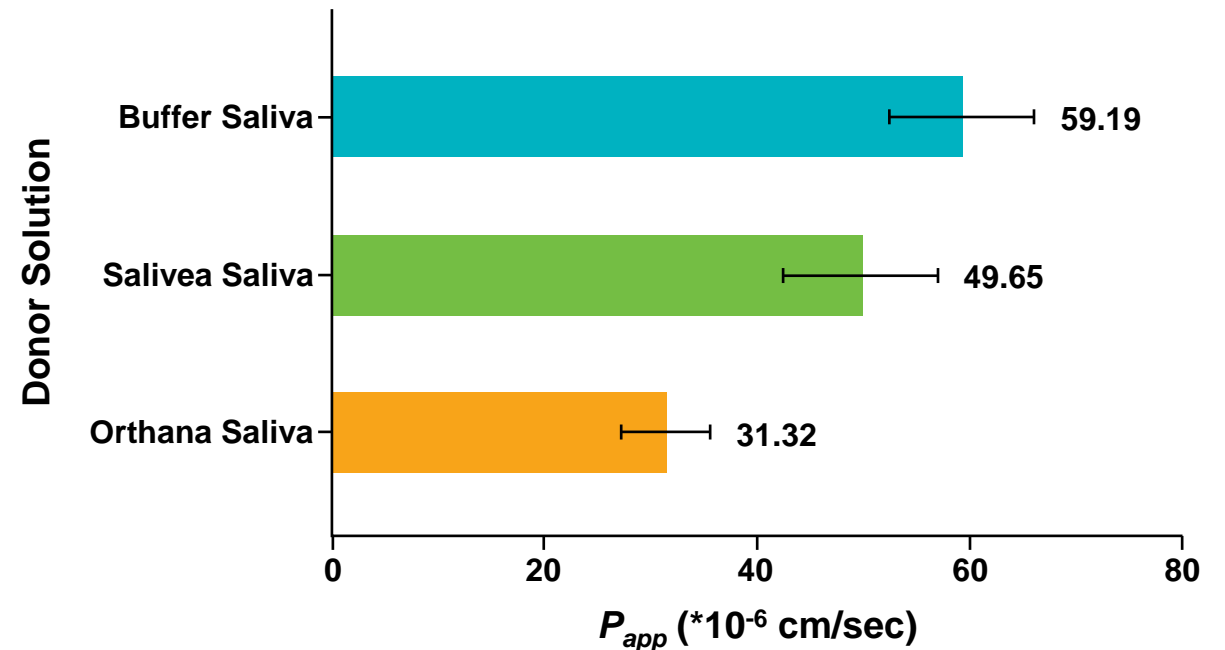
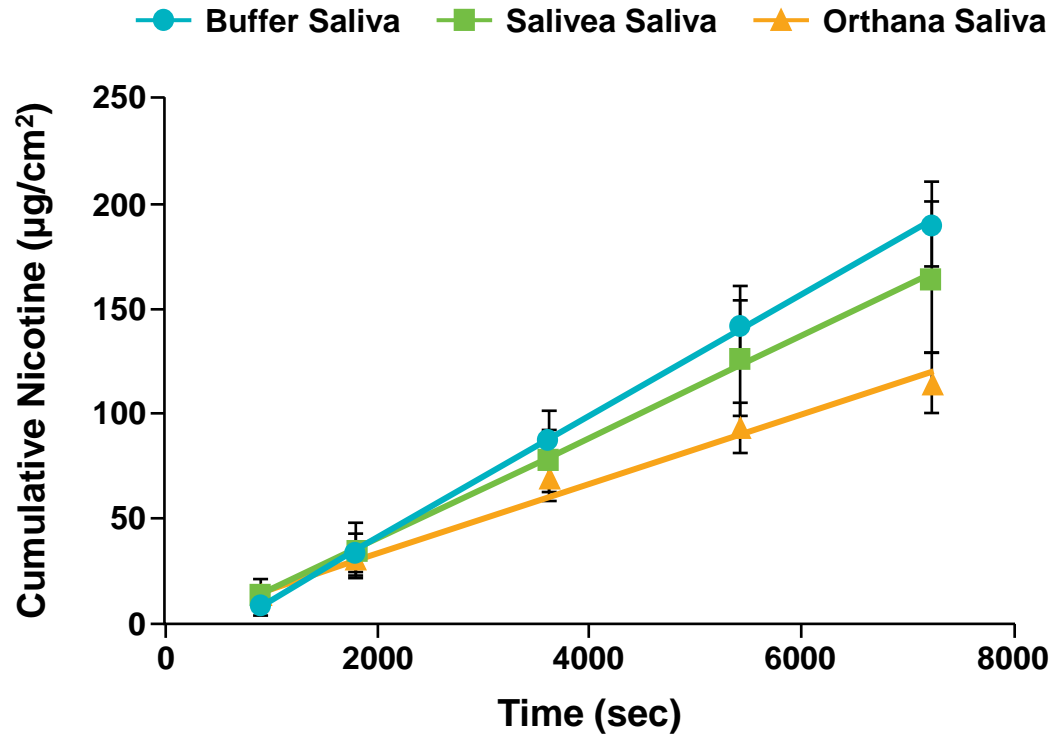


Nicotine permeability is normalized against  $C_0$   
and is not dependent on concentration in the tested range

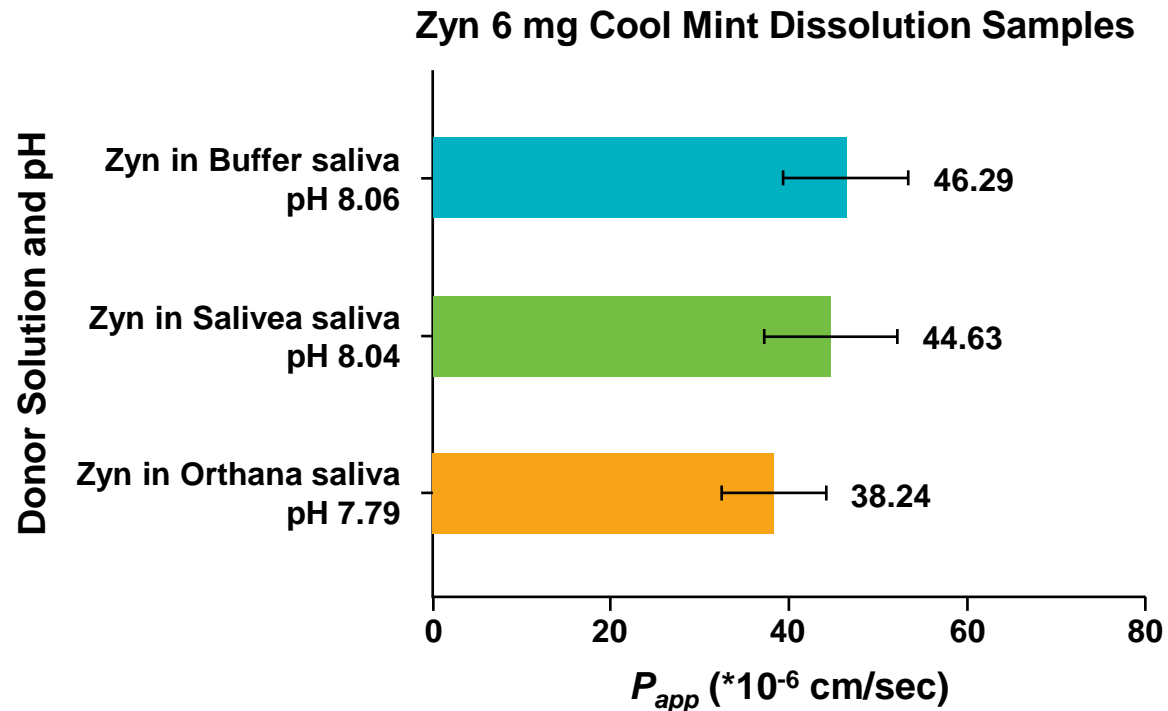


# Artificial Saliva Composition Affects Nicotine Permeability

Donor: 0.5 mg/mL Neat Nicotine Solution in Artificial Saliva, pH 8.5, n=12



# Permeability of Nicotine from *Zyn* 6 mg Cool Mint Dissolution Samples – Comparison of Artificial Saliva



Donor	Extract pH	Conc. of Nicotine (mg/mL)
Buffer saliva	8.06	0.60
Salivea saliva	8.04	0.48
Orthana saliva	7.79	0.60

**The differences in three extracts:**

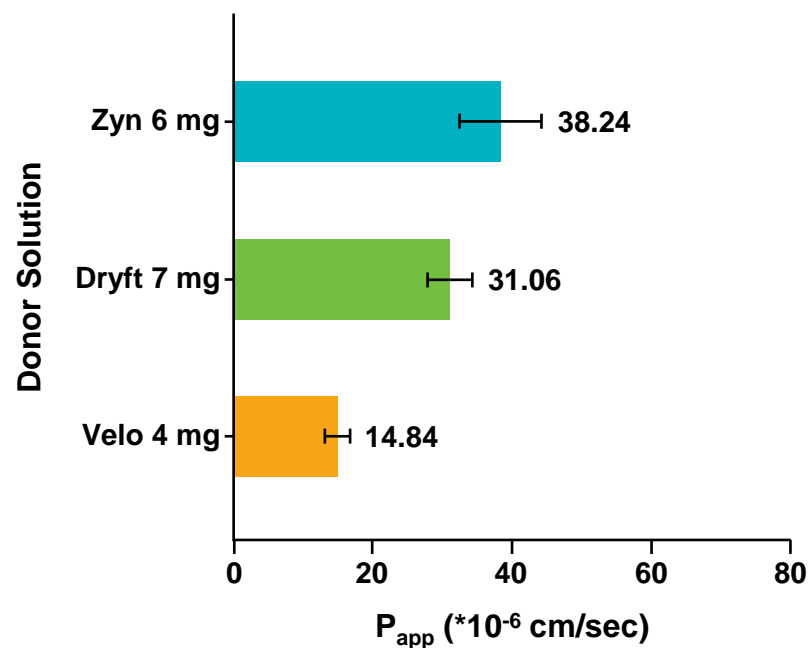
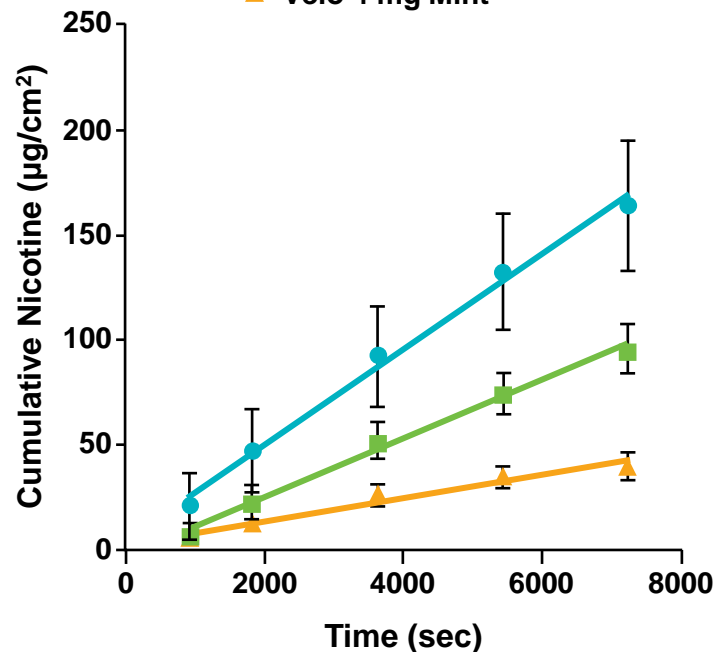
- Nicotine concentration
- pH of the donor solution
- Type of artificial saliva
- Other ingredients released from the product



# Permeability of Nicotine in Orthana Saliva – Comparison of Products

Donor: Dissolution Samples in Orthana Saliva, n=12

● Zyn 6 mg Cool Mint    ■ Dryft 7 mg Spearmint  
▲ Velo 4 mg Mint



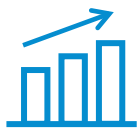
Donor	Extract pH	Conc. of Nicotine (mg/mL)
Zyn 6 mg Cool Mint	7.79	0.60
Dryft 7 mg Spearmint	7.74	0.45
Velo 4 mg Mint	6.94	0.38

**Zyn 6 mg Cool Mint has higher permeability than *Dryft* 7 mg Spearmint at the same pH and in the same solvent**

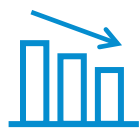




# Conclusion



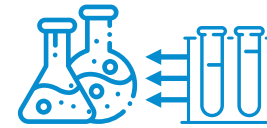
*Zyn* has  
**HIGHER**  
permeability  
than *Dryft*,  
may be due to  
formulation



*Velo* has the  
**LOWEST**  
permeability,  
may be due to  
pH and / or  
formulation



Permeability is  
**INDEPENDENT**  
of nicotine  
concentration in  
the tested range



Permeability  
can be  
**INFLUENCED**  
by pH and artificial  
saliva composition





# Acknowledgements



Shuzhong "Mark"  
Zhuang, Ph. D.

Controlled Release and Delivery  
Technology Group, CIPO,  
**Altria Client Services**,  
Richmond, VA, USA



Fadi Aldeek, Ph. D.

Oral Product Assessment Group,  
Analytical Sciences,  
**Altria Client Services**,  
Richmond, VA, USA





# THANK YOU!

